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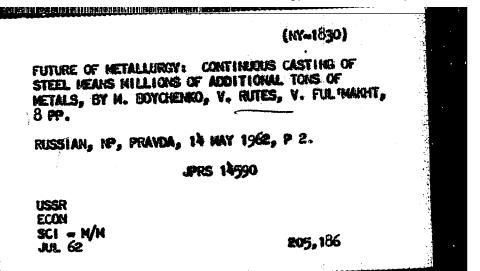
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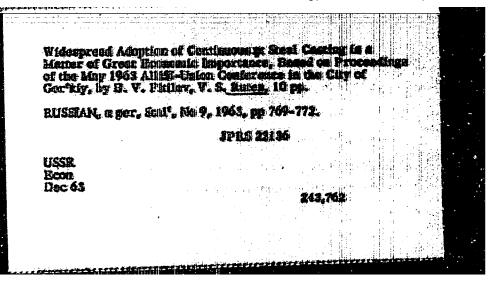
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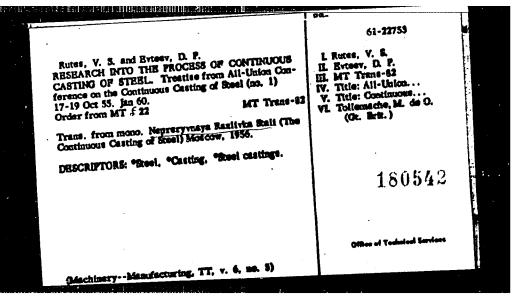
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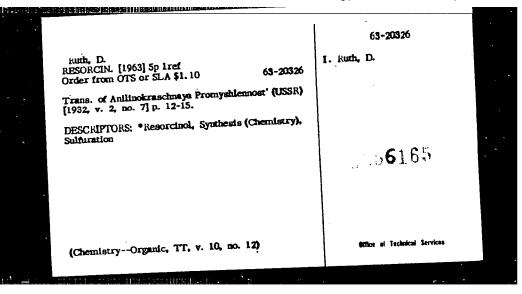
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DESCRIPTORS: \*Pickling, \*Pickling compounds, Recovery, Regeneration, Chlorides, Iron compounds, Chemical reactions, Hydrogen, Gases, Separation, Decomposition, \*Chemical waste.

The process is characterized by reacting with gaseous hydrogen chloride the pickling liquids which are enriched with metallic salts, preferably with iron salts, by separating the precipitated solid metallic chlorides and by decomposing them by means of hydrogen gas into metal and hydrogen chloride, which can again be applied for the precipitation of iron chloride from the pickling liquor.

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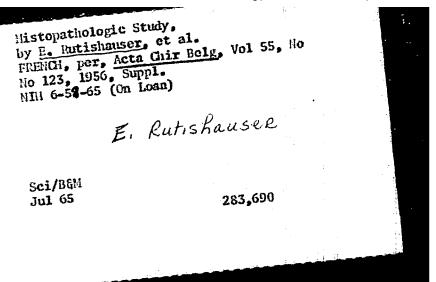
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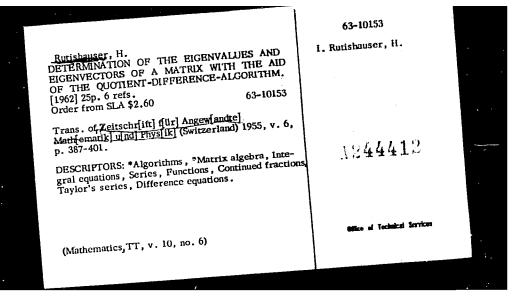
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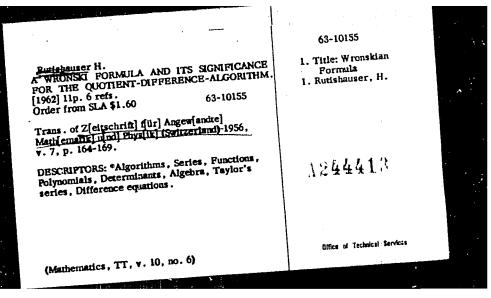
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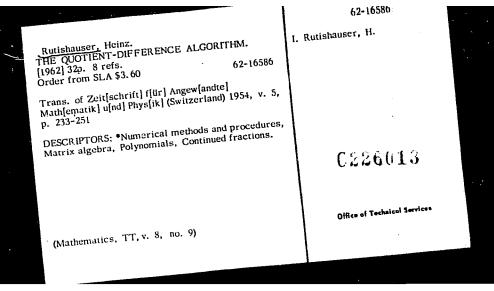
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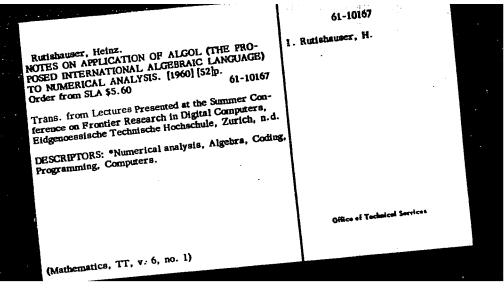
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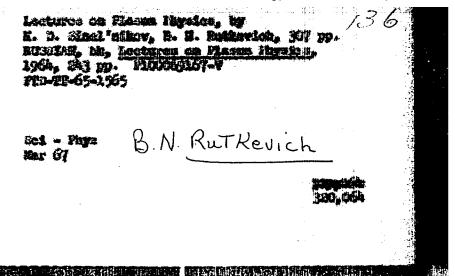
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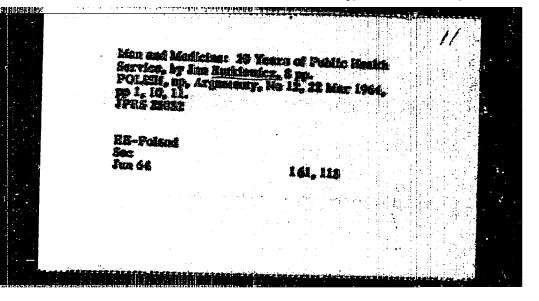
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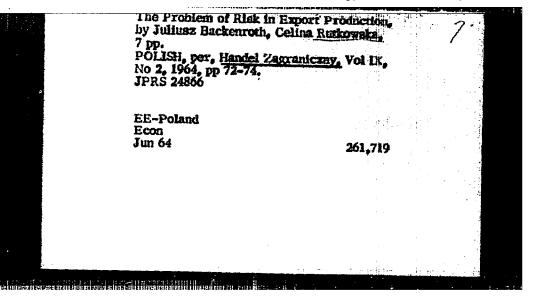
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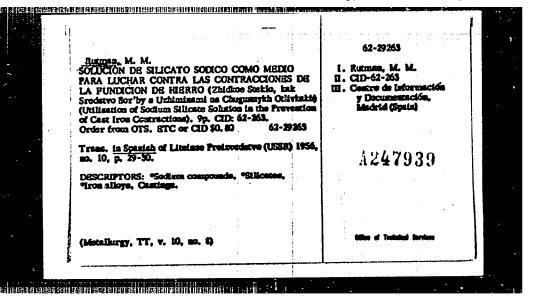
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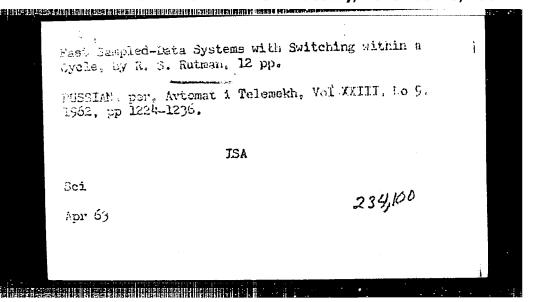
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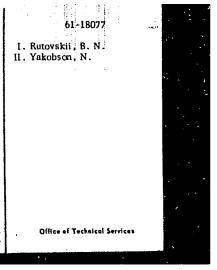
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Rutovskii, B. N. and Yakobson, N. COPOLYMERIZATION OF METHYL ISOPROPENYL KETONE AND METHYL METHACRYLATE. [1961] 7p. 6 refs. Order from OTS or SLA \$1.10

Condensed trans. of Zhurnal Prikladnoi Khimii (USSR) 1941, v. 14, no. 4/5, p. 528-538.

DESCRIPTORS: \*Copolymerization, Ketones, \*Acrylic resins, \*Polymers.

Experiments were carried out showing that copolymers of methyl methacrylate and methyl isopropenyl ketone can be prepared in which the monomers are equally distributed through the individual fractions of the product. In the absence of solvents, copolymers were obtained with a content of the ketone amounting to 25 and 12.5%. In the presence of 50% ketone, no polymer is formed. Copolymerization in solution gave copolymers containing 50.25 and 12.5% ketone. The polymers pre-(Chemistry--Organic, TT, v. 6, no. 7) (over)



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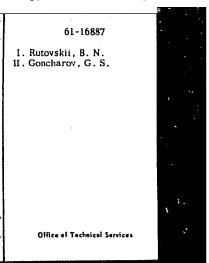
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Condensed trans. of Zhurnal Prikladnoi Khimii (USSR) 1941, v. 14, no. 4/5, p. 542-550.

DESCRIPTORS: \*Copolymerization, \*Styrenes, Ketones, Polymers.

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The method and conditions favoring copolymerization were investigated on the example of styrene and methyl isopropenyl ketone. Small amounts of methyl isopropenyl ketone accelerate polymerization of styrene in emulsions, but additions exceeding 10% of the ketone retard the process. Polymerization in emulsion in the presence of benzoyl peroxide does not give copolymers, while in the presence of hydrogen peroxide, conditions may be created leading to its formation. Under favorable conditions, copolymers are formed the chemical (Chemistry--Organic, TT, v. 6, no. 7) (over)



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